JAIST Repository

https://dspace.jaist.ac.jp/

Title	Excellent uniaxial alignment of poly(9,9- dioctylfluorenyl-2,7-diyl) induced by photoaligned polyimide films
Author(s)	Sakamoto, Kenji; Usami, Kiyoaki; Uehara, Yoichi; Ushioda, Sukekatsu
Citation	Applied Physics Letters, 87(21): 211910-1-211910- 3
Issue Date	2005-11-17
Туре	Journal Article
Text version	publisher
URL	http://hdl.handle.net/10119/4162
Rights	Copyright 2005 American Institute of Physics. This article may be downloaded for personal use only. Any other use requires prior permission of the author and the American Institute of Physics. The following article appeared in Sakamoto, K., Usami, K., Uehara, Y., Ushioda, S., Applied Physics Letters 87(21), 211910 (2005) and may be found at http://link.aip.org/link/?APPLAB/87/211910/1 .
Description	



Excellent uniaxial alignment of poly(9,9-dioctylfluorenyl-2,7-diyl) induced by photoaligned polyimide films

Kenji Sakamoto^{a)}

Nanomaterials Laboratory, National Institute for Materials Science, 1-1 Namiki, Tsukuba, Ibaraki 305-0044, Japan

Kiyoaki Usami

RIKEN Photodynamics Research Center, 519-1399 Aramaki-aoba, Aoba-ku, Sendai 980-0845, Japan and Nanomaterials Laboratory, National Institute for Materials Science, 1-1 Namiki, Tsukuba, Ibaraki 305-0044, Japan

Yoichi Uehara

Research Institute of Electrical Communication, Tohoku University, Sendai 980-8577, Japan and RIKEN Photodynamics Research Center, 519-1399 Aramaki-aoba, Aoba-ku, Sendai 980-0845, Japan

Sukekatsu Ushioda

Japan Advanced Institute of Science and Technology, 1-1 Asahidai, Nomi, Ishikawa 923-1292, Japan and RIKEN Photodynamics Research Center, 519-1399 Aramaki-aoba, Aoba-ku, Sendai 980-0845, Japan

(Received 10 May 2005; accepted 29 September 2005; published online 17 November 2005)

We have investigated the alignment of poly (9,9-dioctylfluorenyl-2,7-diyl) (PFO) induced by photoaligned polyimide films. To induce anisotropic orientation of polyimide backbone structures by optical treatment, we used a specially designed polyimide (Azo-PI), which contains azobenzene in the backbone structure. The 30-nm-thick PFO layer, spincoated onto a photoaligned Azo-PI film, was heated to the liquid crystalline phase of PFO, and then cooled down to room temperature at $-0.1 \,^{\circ}C/min$. The thermally treated PFO layer showed a polarization ratio of approximately 30 in photoluminescence. This high polarization ratio has not previously been reported in PFO films formed by other alignment techniques. This result indicates that the photoaligned Azo-PI film has an excellent ability to align liquid crystalline polymers. © 2005 American Institute of Physics. [DOI: 10.1063/1.2135873]

Alignment control of conjugated polymers is of great importance for the fabrication of polymer-based electronic devices with high performance. The charge carrier mobility of conjugated polymer films can be significantly enhanced by aligning the polymer backbone structures along a certain direction.¹ This enhancement improves the switching frequency of polymer-based thin-film transistors (TFTs). Polarized light-emitting diodes (LEDs) can be realized by uniaxially aligning light-emitting polymers. Uniaxial alignment of conjugated polymers is achieved by various techniques, such as stretching,² direct rubbing,³ the Langmuir–Blodgett technique,⁴ and a friction transfer method.⁵

An attractive way to realize uniaxial alignment is to use the liquid-crystalline nature of polymers, e.g., polyfluorene (PF) derivatives,^{6,7} which have rigid backbone structures with flexible side chains. By depositing liquid-crystalline polymers onto a proper alignment layer and subsequently performing a suitable thermal treatment, uniaxial alignment of the polymer can be obtained. Rubbed films of polyimide^{1,7–9} and poly (p-phenylenevinylene) (PPV) (Ref. 10) are often used as the alignment layer. However, mechanical rubbing produces dust and scratches, and its uniformity is poor.¹¹ These drawbacks are expected to severely affect the alignment of conjugated polymers, especially in thin films. Thus, the development of a noncontact alignment technique is of crucial importance for improvements in the device performance. Photoalignment is a promising noncontact alternative to the mechanical rubbing technique. Various photoalignment methods^{12–15} have been proposed for low-molecular-weight liquid-crystal molecules. Recently, azobenzene side chain polymers have been used as photoalignment layer materials to induce uniaxial alignment of PF.¹⁶ However, the photoinduced alignment of the azobenzene side chains is not permanent and can be reoriented by subsequent irradiation of light.¹⁷ Unexpected reorientation of the azobenzene side chain might cause the deterioration of the alignment of conjugated polymers in the overlying layer. Stable photoalignment layers are required for practical applications.

We are focusing on a method using the photoisomerization reaction of a polyamic acid containing azobenzene in the backbone structure (Azo-PAA).^{14,18} As described below, the alignment layer obtained by this method is polyimide, which is chemically and thermally stable.¹⁹ The photoinduced alignment of the Azo-PAA backbone structure is stabilized by thermal imidization.^{14,20} Thus, this photoalignment method provides optically, chemically, and thermally stable alignment layers. In this letter, we have shown that the photoalignment layer has an excellent ability to align PF, which is an attractive material for polymer-based LEDs and TFTs.

The molecular structure of Azo-PAA used in this study is shown in Fig. 1(a). The Azo-PAA films were spincoated onto quartz substrates, and then irradiated with linearly polarized light (LP-L) at normal incidence. A 500 W deep ultraviolet (UV) lamp (Ushio Inc. UXM-501MD) was used as the light source, and the wavelength was selected with a band-pass filter of transmission wavelength 340–500 nm (Asahi Spec-

0003-6951/2005/87(21)/211910/3/\$22.50

87. 211910-1

Downloaded 25 Feb 2008 to 150.65.7.70. Redistribution subject to AIP license or copyright; see http://apl.aip.org/apl/copyright.jsp

^{a)}Electronic mail: sakamoto.kenji@nims.go.jp

^{© 2005} American Institute of Physics



FIG. 1. The molecular structure of (a) Azo-PAA, (b) Azo-PI, and (c) PFO used in this study.

tra Co., Ltd.). To produce linear polarization, a Glan-Taylor prism polarizer was used. The LP-L exposure was varied up to 155 J/cm² to induce different surface anisotropies. After the photoalignment treatment, the Azo-PAA films were converted into the corresponding polyimide (Azo-PI) films by thermal imidization at 250 °C for 2 h in a nitrogen atmosphere. The molecular structure of Azo-PI is shown in Fig. 1(b). The thickness of the Azo-PI films was ~12 nm. The molecular orientation was determined by measuring the polarized UV-visible (UV-VIS) absorption spectra at normal incidence.

(9.9 -The PF used in this study was poly dioctylfluorenyl-2,7-diyl) 3,5endcapped with dimethylphenyl (PFO), which was used as received from American Dye Source, Inc. Its molecular structure is shown in Fig. 1(c). Approximately 30 nm thick PFO layers were deposited on the Azo-PI films by spincoating a filtered toluene solution of PFO (1 wt %) at 3000 rpm for 60 s. To align the PFO, the samples were annealed at 220 °C (liquidcrystalline phase of PFO)²¹ for 2 h, and then cooled down to room temperature at -0.1 °C/min, which formed crystalline PFO layers.²¹ The thermal treatment was performed in a nitrogen atmosphere to prevent thermal oxidation. The molecular orientation of the PFO layers was determined by measuring the polarized photoluminescence (PL) spectra with a fluorescence spectrometer (JASCO FP777). Unpolarized light of wavelength 390 nm was used as the excitation light, which impinged onto the PFO layer at 45° from the surface normal. The light emitted from the PFO layer was collected along the surface normal through a Glan-Thompson prism polarizer.

Figure 2(a) shows the polarized UV-VIS absorption



FIG. 2. (a) Polarized UV-VIS absorption spectra of the Azo-PI film that was obtained by thermally imidizing an Azo-PAA film exposed to LP-L of 39 J/cm². A_{\perp} and A_{\parallel} are the spectra for the light polarized perpendicular and parallel, respectively, to the polarization direction of the LP-L. (b) Polarized PL spectra of the crystalline PFO layer formed on the same Azo-PI film. I_{\perp} and I_{\parallel} are the perpendicular and parallel component spectra with respect to the polarization direction of the LP-L.



FIG. 3. LP-L exposure dependence of (a) the dichroic ratio $(A_{\perp}/A_{\parallel} \text{ at } 351 \text{ nm})$ of the Azo-PI films and (b) the PL polarization ratio $(I_{\perp}/I_{\parallel} \text{ at } 432 \text{ nm})$ of the crystalline PFO layer. The horizontal straight line shows the polarization ratio of 30. The inset in (b) shows the LP-L dependence of I_{\perp}/I_{\parallel} in the low LP-L exposure region.

spectra of the Azo-PI film that was obtained by thermal imidization of the Azo-PAA film exposed to LP-L of 39 J/cm². Here, A_{\perp} and A_{\parallel} are absorbance for the light polarized perpendicular and parallel, respectively, to the polarization direction of the LP-L. A well-defined peak was observed at 351 nm, which is assigned to the π - π^* transition of azobenzene in the Azo-PI backbone structure.²² The dichroic ratio defined by A_{\perp}/A_{\parallel} was 2.1. Since the transition dipole lies parallel to the backbone structure, the dichroic ratio reflects the anisotropic orientation of the Azo-PI backbone structure. Thus, we see that the Azo-PI backbone structures align on average perpendicular to the polarization direction of the LP-L.

Figure 2(b) shows the polarized PL spectra of the crystalline PFO layer formed on the Azo-PI film (LP-L exposure: 39 J/cm²). I_{\perp} and I_{\parallel} are the perpendicular and parallel components of the PL intensity with respect to the polarization direction of the LP-L. The sharp peaks at 432, 458, 491, and 525 nm are assigned to the excitonic emission and its vibronic progression from noninteracting single chains.^{6,21} Although for crystalline PFO layers the broad fluorescence assigned to emission from excimers or aggregates was often observed in the longer-wavelength region beyond 480 nm,^{5,6,21} it was strongly suppressed for our samples. We observed highly polarized emission from the PFO layer, whose polarization ratio defined by I_{\perp}/I_{\parallel} was 30 at 432 nm. This polarization ratio corresponds to the uniaxial molecular order parameter of $0.91[=(I_{\perp}-I_{\parallel})/(I_{\perp}+2I_{\parallel})]$. To our knowledge, the highest polarization ratio in PL reported previously was 25, which was observed for the crystalline PFO layer prepared by a friction-transfer method followed by a slowcooling thermal treatment.⁵ The highly polarized PL observed in this study indicates that the photoaligned Azo-PI film is an excellent alignment layer for PFO.

Figures 3(a) and 3(b) show the LP-L exposure dependence of the dichroic ratio A_{\perp}/A_{\parallel} of the Azo-PI films and the polarization ratio I_{\perp}/I_{\parallel} of the crystalline PFO layers, respectively. The dichroic ratio of the Azo-PI film increases monotonically with the LP-L exposure, although the rate of increase decreases as the LP-L exposure increases. It reaches 3.3 at 155 J/cm². On the other hand, the polarization ratio of the PFO layer increases rapidly up to ~30, and beyond 2.8 J/cm² it becomes constant. This result shows that 2.8 J/cm² is enough LP-L exposure to realize the highly oriented PFO layer.

Downloaded 25 Feb 2008 to 150.65.7.70. Redistribution subject to AIP license or copyright; see http://apl.aip.org/apl/copyright.jsp

Before we proceed to further discussion, we would like to mention the relationship between the in-plane anisotropies of the Azo-PI film and its surface. In this study, we used 12 nm thick Azo-PI films as photoalignment layers. For such very thin films, we can assume that the photoaligned Azo-PI film has a uniform in-plane molecular orientation over the entire film thickness.¹⁸ Thus, the dichroic ratio shown in Fig. 3(a) reflects not only the in-plane anisotropy of the Azo-PI film but also that of its surface. Indeed, we have recently reported that the in-plane anisotropy of a 15 nm thick photoaligned Azo-PI film is transferred to the 4-n-octyl-4'-cyanobiphenyl monolayer in contact with it.23 This experimental result strongly supports the above assumption. Therefore, we believe that A_{\perp}/A_{\parallel} shown in Fig. 3(a) can be considered as the in-plane anisotropy of the Azo-PI film surface.

Now we can discuss the relationship between the inplane anisotropy of the Azo-PI film surface $(A_{\perp}/A_{\parallel})$ and the molecular order of the crystalline PFO layer $(I_{\perp}/I_{\parallel})$. From Fig. 3, we see that the molecular order of the PFO layer is much higher than the in-plane anisotropy of the Azo-PI film surface. This result suggests that the molecular order of the crystalline PFO layer is determined by the bulk nature of PFO, rather than the interaction between the Azo-PI and PFO molecules. Here, the bulk nature of PFO means the selforganization of PFO in the liquid-crystalline phase and in the crystallization process from the liquid-crystalline phase. This is supported by the fact that for $A_{\parallel}/A_{\parallel} \ge 1.25$, the PL polarization ratio of the PFO layer is ~ 30 , independent of A_{\perp}/A_{\parallel} . The in-plane anisotropy of the Azo-PI film surface does not affect the molecular order of the crystalline PFO layer when the in-plane anisotropy is sufficiently large $(A_{\perp}/A_{\parallel} \ge 1.25)$. The main role of the Azo-PI film is to determine the preferential alignment direction of PFO.

The formation of highly oriented PFO layers can be explained as follows. When the PFO layer is heated to the liquid-crystalline phase, the PFO molecules self-align along the preferential alignment direction determined by the underlying Azo-PI film. At this stage, the PL polarization ratio increases up to $\sim 10^{.10,24}$ The subsequent crystallization of the PFO layer from the liquid-crystalline phase further enhances the orientational molecular order, and the PL polarization ratio ratio reaches ~ 30 .

When the in-plane anisotropy of the Azo-PI film surface is small $(A_{\perp}/A_{\parallel} < 1.25)$, the small in-plane anisotropy causes lowering of the molecular order of the PFO layer as seen in Fig. 3. This is due to the insufficient surface anchoring strength of the Azo-PI film along the preferential alignment direction. Thus the photoaligned Azo-PI film with $A_{\perp}/A_{\parallel} \ge 1.25$ must be used to obtain a highly oriented crystalline PFO layer.

To conclude, we have investigated the molecular orientation of the crystalline PFO layers induced by photoaligned Azo-PI films, which are thermally, chemically, and optically stable. We succeeded in forming highly oriented PFO layers that showed a polarization ratio of approximately 30 in PL. This polarization ratio corresponds to a uniaxial molecular order parameter of 0.91. To our knowledge, the polarization ratio of \sim 30 has not previously been reported in PFO films formed by other alignment techniques. This means that the photoaligned Azo-PI film is an excellent alignment layer for PFO derivatives and probably for other liquid-crystalline polymers. We believe that the excellent alignment ability is attributed to the uniformity and the sufficiently large in-plane anisotropy of the photoaligned film surface.

The PL measurements were performed at the Photonics Research Institute of the National Institute of Advanced Industrial Science and Technology (AIST). The authors are grateful to J. Nishizawa for his valuable advice. They would like to thank K. Yase, Y. Yoshida, S. Nagamatsu, and M. Misaki of AIST for their valuable comments and support of the PL measurements. They would also like to thank H. Naito of Osaka Prefecture University for his helpful comments on the sample preparation. They thank S. Murata, N. Narita, and J. Yokota of Chisso Co. Ltd. for supplying the Azo-PAA solution used in this study, K. Miki for his valuable discussion, and J. H. G. Owen for his helpful comments on the manuscript.

- ¹H. Sirringhaus, R. J. Wilson, R. H. Friend, M. Inbasekaran, W. Wu, E. P.
- Woo, M. Grell, and D. D. C. Bradley, Appl. Phys. Lett. 77, 406 (2000).
- ²C. Weder, C. Sarwa, A. Montali, C. Bastiaansen, and P. Smith, Science **279**, 835 (1998).
- ³M. Hamaguchi and K. Yoshino, Appl. Phys. Lett. 67, 3381 (1995).
- ⁴V. Cimrova, M. Remmers, D. Neher, and G. Wegner, Adv. Mater. (Weinheim, Ger.) **8**, 146 (1996).
- ⁵M. Misaki, Y. Ueda, S. Nagamatsu, Y. Yoshida, N. Tanigaki, and K. Yase, Macromolecules **37**, 6926 (2004).
- ⁶D. Neher, Macromol. Rapid Commun. **22**, 1366 (2001).
- ⁷M. Grell, D. D. C. Bradley, M. Inbasekaran, and E. P. Woo, Adv. Mater. (Weinheim, Ger.) **9**, 798 (1997).
- ⁸H. Sirringhaus, T. Kawase, R. H. Friend, T. Shimoda, M. Inbasekaran, W. Wu, and E. P. Woo, Science **290**, 2123 (2000).
- ⁹A. Kobayashi, M. Todorokihara, K. Fujihara, and H. Naito, Synth. Met. **135**, 295 (2003).
- ¹⁰K. S. Whitehead, M. Grell, D. D. C. Bradley, M. Jandke, and P. Strohriegl, Appl. Phys. Lett. **76**, 2946 (2000).
- ¹¹D.-H. Chung, Y. Takanishi, K. Ishikawa, H. Takezoe, and B. Park, Jpn. J. Appl. Phys., Part 1 **40**, 1342 (2001).
- ¹²P. O. Jackson, M. O'Neill, W. L. Duffy, P. Hindmarsh, S. M. Kelly, and G. J. Owen, Chem. Mater. **13**, 694 (2001).
- ¹³W. M. Gibbons, P. J. Shannon, S.-T. Sun, and B. J. Swetlin, Nature (London) **351**, 49 (1991).
- ¹⁴B. Park, Y. Jung, H.-H. Choi, H.-K. Hwang, Y. Kim, S. Lee, S.-H. Jang, M.-A. Kakimoto, and H. Takezoe, Jpn. J. Appl. Phys., Part 1 **37**, 5663 (1998).
- ¹⁵M. Hasegawa and Y. Taira, J. Photopolym. Sci. Technol. 8, 241 (1995).
- ¹⁶X. H. Yang, D. Neher, S. Lucht, H. Nothofer, R. Güntner, U. Scherf, R. Hagen, and S. Kostromine, Appl. Phys. Lett. **81**, 2319 (2002).
- ¹⁷D. Sainova, A. Zen, H.-G. Nothofer, U. Asawapirom, U. Scherf, R. Hagen, T. Bieringer, S. Kostromine, and D. Neher, Adv. Funct. Mater. **12**, 49 (2002).
- ¹⁸K. Sakamoto, K. Usami, M. Kikegawa, and S. Ushioda, J. Appl. Phys. **93**, 1039 (2003).
- ¹⁹C. E. Sroog, J. Polym. Sci. A **11**, 161 (1976).
- ²⁰K. Usami, K. Sakamoto, Y. Uehara, and S. Ushioda, Jpn. J. Appl. Phys., Part 1 44, 6703 (2005).
- ²¹B. Schartel, V. Wachtendorf, M. Grell, D. D. C. Bradley, and M. Hennecke, Phys. Rev. B 60, 277 (1999).
- ²²H. Fliegl, A. Köhn, C. Hättig, and R. Ahlrichs, J. Am. Chem. Soc. **125**, 9821 (2003).
- ²³K. Usami, K. Sakamoto, Y. Uehara, and S. Ushioda, Appl. Phys. Lett. 86, 211906 (2005).
- ²⁴K. Sakamoto, K. Usami, Y. Uehara, and S. Ushioda, Colloids Surf., A (to be published).